

NMR determination of the structure of Julibroside J₁

Libin Ma ^{a,b,*}, Guangzhong Tu ^a, Siping Chen ^c, Ruyi Zhang ^c,
Luhua Lai ^b, Xiaojie Xu ^b, Youqi Tang ^b

^a Beijing Institute of Microchemistry, Beijing 100091, People's Republic of China

^b Department of Chemistry, Peking University, Beijing 100871, People's Republic of China

^c Beijing Medical University, Beijing 100083, People's Republic of China

Received 9 March 1995; accepted 1 September 1995

Abstract

Julibroside J₁ is a new triterpenoid saponin which contains one triterpene, two monoterpenes and nine sugar residues. The structure has been determined almost exclusively by high-resolution NMR methods. The ¹H and ¹³C NMR spectra of Julibroside J₁ C₅D₅N have been assigned by homonuclear and heteronuclear correlation experiments, such as COSY, CH-COSY, TOCSY, HMBC, HMQC-COSY, HMQC-TOCSY and NOESY. Anomeric configurations were obtained by combined use of ¹J_{CH} and ³J_{H1,H2} and NOESY data. The particular sugar residues were identified by utilizing ³J_{HH} values obtained from TOCSY cross-peaks, NOE difference spectra, and several 1D-TOCSY spectra, and three-bond intra-ring cross-peaks from a HMBC spectrum. Linkage assignments were made using the HMBC spectrum, and supplemented by NOE data from the NOESY spectrum. The structure of Julibroside J₁ was characterized as 3-O-[β-D-xylopyranosyl-(1 → 2)-α-L-arabinopyranosyl-(1 → 6)-β-D-glucopyranosyl]-21-O-[(6S)-2-trans-2-hydroxymethyl-6-methyl-6-O-[4-O-((6S)-2-trans-2,6-dimethyl-6-O-(6-deoxy-β-D-glucopyranosyl)-2,7-octadienoyl)-6-deoxy-β-D-glucopyranosyl]-2,7-octadienoyl]-acacic acid 28-O-[β-D-glucopyranosyl-(1 → 3)-[α-L-arabinofuranosyl-(1 → 4)]-α-L-rhamnopyranosyl-(1 → 2)]-β-D-glucopyranosyl ester.

Keywords: Julibroside; NMR; HMQC-TOCSY

1. Introduction

Julibroside J₁ (Scheme 1) is an important drug extracted from the dried stem bark of the silk tree *Albizia julibrissin* DURAZZ (leguminosol), Albizziac Cortex used as a

* Corresponding author.

Table 1
 ^{13}C NMR data (δ in ppm)^a for the triterpene moiety of J_1

Carbon	J_1	Ref. [7]	Carbon	J_1	Ref. [7]
1	38.9	39.0	16	74.8	74.4
2	26.9	26.8	17	51.7	51.8
3	88.9	88.1	18	40.9	41.1
4	39.7	39.6	19	47.9	48.6
5	56.1	56.0	20	35.6	35.4 ^b
6	18.7	18.9	21	77.1	77.0 ^b
7	33.7	33.6	22	36.4	36.5 ^b
8	40.2	39.9	23	28.3	28.8
9	47.2	47.3	24	17.1	15.1
10	37.1	37.5	25	15.9	16.6
11	23.9	23.9	26	17.3	17.6
12	124.1	122.7	27	27.3	27.6
13	143.3	144.5	28	174.4	174.8 ^b
14	43.0	42.1	29	29.3	29.2 ^b
15	35.9	35.9	30	19.1	19.2 ^b

^a Chemical shifts with the highest field signal of $\text{C}_5\text{D}_5\text{N}$ (δ 123.5 ppm) as reference.

^b Data from ref. [4].

traditional Chinese drug. In the literature [1–8], structures of triterpenoid saponins have been characterized utilizing chemical methods combined with ^1H and ^{13}C NMR spectroscopy. For the characterization of complex carbohydrates, the development of 2D-NMR techniques enables the complete assignment of ^1H and ^{13}C NMR spectra [9]. In addition, NMR spectroscopy is a non-destructive method that requires just a few milligrams of compound to allow a complete analysis to be performed [10]. This is an attractive alternative to chemical methods which are more tedious and subject to complications or failure.

2. Results and discussion

Identification of triterpene and two monoterpenes.—The ^1H and ^{13}C signal assignments of the triterpene [7] and two monoterpenes [3] of Julibroside J_1 were obtained by comparison with the ^{13}C data of similar structures in the literature, and subsequently confirmed by the data from COSY, TOCSY, CH-COSY and HMBC spectra. The results are summarized in Tables 1–3.

Identification and NMR signal assignments of sugar residues.—Individual sugar residues are indicated by capital letters (A–I, see Scheme 1) in the structure of J_1 , and ^{13}C resonances are identified by a capital letter and number of the ring carbon atom. The ^{13}C NMR spectrum gave nine resolved anomeric ^{13}C resonances at 101.76 (A_1), 99.29 (B_1), 99.19 (C_1), 95.67 (D_1), 106.21 (E_1), 102.22 (F_1), 105.73 (G_1), 106.74 (H_1) and 111.02 (I_1) ppm. Therefore, the proton resonances at 5.880, 4.380, 4.815, 6.030, 4.980, 5.140, 5.310, 4.880, and 6.250 ppm were easily assigned by direct correlations from the CH-COSY spectrum (data not shown) as the anomeric protons of residues A, B, C, D,

Table 2
¹H NMR data (δ in ppm)^a for the triterpene moiety of J₁

Proton ^b	δ	Proton	δ	Proton	δ
1α	1.15	9α	1.89	21α	6.29
1β	1.65	11a	2.07	22α	2.19
2α	1.92	11b	2.07	22β	2.72
2β	2.30	12	5.60	23α	1.29
3α	3.49	15α	2.03	24β	1.01
5α	0.93	15β	2.23	25β	0.96
6a	1.48	16β	5.21	26β	1.16
6b	1.48	18α	3.42	27α	1.87
7a	1.72	19α	1.44	29β	1.04
7b	1.72	19β	2.94	30α	1.09

^a Chemical shifts with the highest field signal of C₅D₅N (δ 7.19 ppm) as reference.

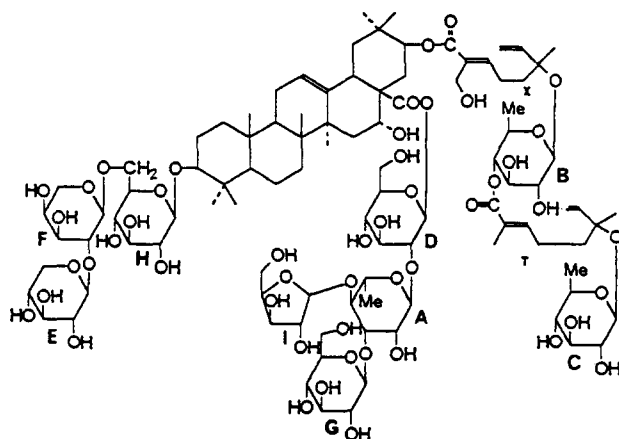
^b Stereospecific assignment was obtained from NOE data.

E, F, G, H, and I, respectively. ¹³C signals at 61.95 (D₆), 67.16 (E₅), 64.20 (F₅), 62.76 (G₆), 69.52 (H₆) and 62.55 (I₅) ppm were identified as methylene carbons from the DEPT spectrum (data not shown), which also shows three methyl carbon signals at 18.81 (A₆), 17.09 (B₆) and 18.81 (C₆) ppm. Similarly, in the CH-COSY spectrum, the three methyl carbon resonances show correlations with methyl proton resonances at 1.760, (H-A₆), 1.340 (H-B₆) and 1.590 (H-C₆) ppm and six methylene ¹³C resonances show correlations with their proton resonances at 4.315, 4.190 (H-D₆'s), 4.390, 4.590 (H-E₅'s), 4.290, 3.740 (H-F₅'s), 4.480, 4.180 (H-G₆'s), 4.630, 4.220 (H-H₆'s) and 4.240, 4.130 (H-I₅'s) ppm, respectively. The assignments of methylene proton resonances were confirmed by the COSY spectrum (data not shown) in which intensive

Table 3
 NMR data (δ in ppm)^a for two monoterpenes of J₁

No.	T			X		
	δ _H	δ _C		δ _H	δ _C	
		J ₁	Ref. [3]		J ₁	Ref. [3]
1		167.8	167.2		167.5	167.0
2		127.9	127.6		134.9	132.8
3	7.100	143.5	144.4	7.040	145.2	146.8
4	2.510	23.7	24.0	2.670	23.7	24.1
5	1.805	38.6	41.4	1.800	40.9	41.8
6		79.5	72.2		79.8	72.2
7	6.300	144.3	146.3	6.190	143.9	146.3
8a	5.180	114.3	111.8	5.200	115.2	111.8
8b	5.320			5.400		
9	1.930	12.7	12.6	4.710	56.2	56.2
10	1.450	24.9	28.4	1.500	23.8	28.3

^a Chemical shifts with the highest field signals of C₅D₅N as reference (7.19 ppm for ¹H and 123.5 ppm for ¹³C).

Scheme 1. Structure of Julibroside J_1 .

cross-peaks were observed between geminal proton resonances. The proton and carbon resonances in the middle of rings were then identified by the correlations with anomeric, methylene and methyl protons using COSY, TOCSY, CH-COSY, HMQC-COSY, HMQC-TOCSY and HMBC spectra as discussed below.

Residue G.—In the TOCSY spectrum ($\tau_m = 120$ ms, Fig. 1), the anomeric proton of residue G shows six cross-peaks to each proton in the residue including the methylene proton H-G₆'s. Because of severe overlap in the region of 4.2–3.9 ppm, only H-G₂ (3.970 ppm) and H-G₅ (4.940 ppm) could be traced in the COSY spectrum by cross-peaks H-G₁/H-G₂ and H-G₅/H-G₆. The assignment of H-G₃ and H-G₄ was obtained from a TOCSY spectrum obtained with a shorter spin-lock time ($\tau_m = 60$ ms, data not shown) in which the anomeric proton showed a larger cross-peak to H-G₃ (4.130 ppm) than to H-G₄ (4.055 ppm). The assignment was supported by data from the NOESY spectrum (data not shown) in which the anomeric proton showed cross-peaks to H-G₃ and H-G₅. This information also indicated that the anomeric proton of residue G has a β -configuration [11], consistent with $^1J_{CH}$ (160 Hz) and $^3J_{H1,H2}$ (7.7 Hz) values. In general, $^1J_{CH} < 165$ Hz and $^3J_{H1,H2} > 5$ Hz for β -configurations, and $^1J_{CH} > 165$ Hz and $^3J_{H1,H2} < 5$ Hz for α -configurations [12,13]. In the NOESY spectrum cross-peaks from H-1 to H-3 and H-5 indicated a β -anomer, and a cross-peak from H-1 to H-2 rather than to H-3 and H-5 indicates an α -anomer [11]. *Gluco*, *galacto* and *manno* configurations can be distinguished from TOCSY spectra with relatively short spin-lock times ($\tau_m = 60$ ms) [9] and from the magnitudes of $^3J_{HH}$ [13]. Glucopyranoside has large vicinal I_H – I_H couplings among ring protons due to *trans* diaxial orientations, and correlation peaks can be traced from H-1 to H-5 or H-6 in the TOCSY spectrum ($\tau_m = 60$ ms). Galactopyranoside has a small $^3J_{H4,H5}$, so connectivities can only be traced to H-4 in the TOCSY spectrum ($\tau_m = 60$ ms). For manno- or rhamno-pyranosides, connectivities can be traced only to H-2 in the TOCSY spectrum ($\tau_m = 60$ ms) due to the small value of $^3J_{H1,H2}$. Thus, we assigned residue G to a β -glucopyranoside because it showed correlation peaks from the anomeric proton to H-G₅ in the TOCSY spectrum ($\tau_m = 60$

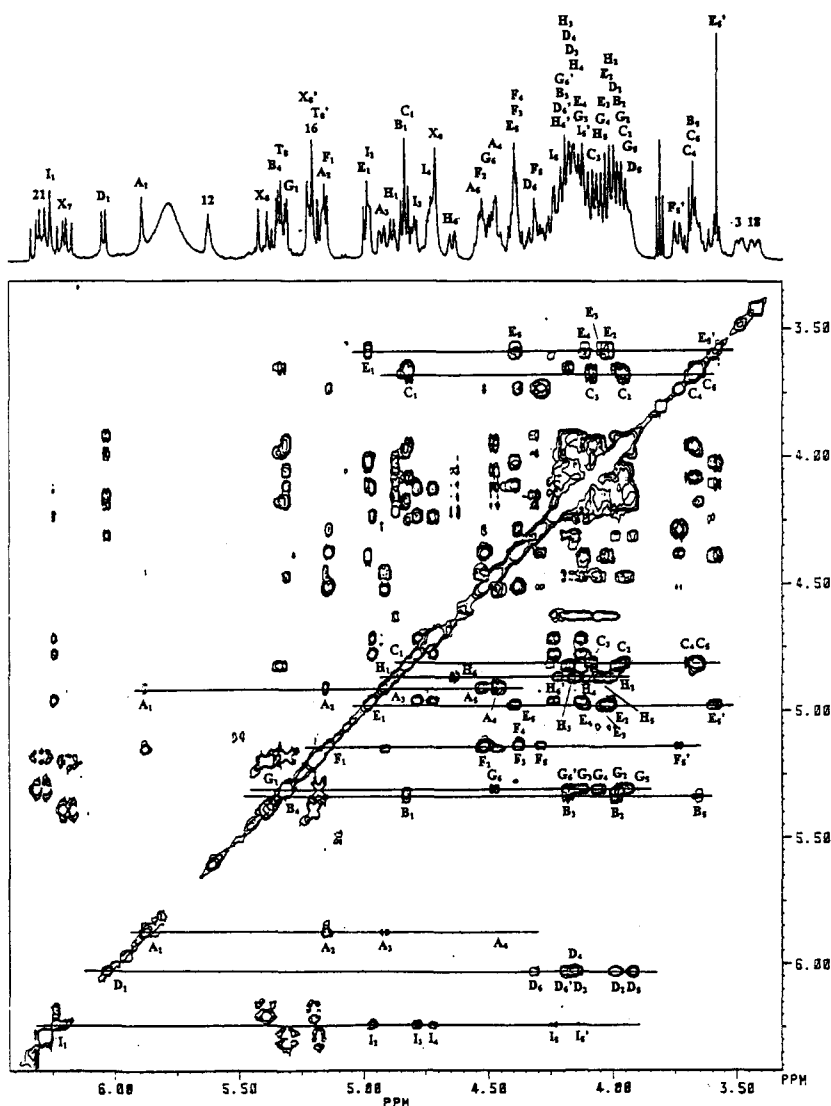


Fig. 1. TOCSY spectrum ($\tau_m = 120$ ms) of the region 6.4–3.3 ppm for Julibroside J_1 , 2 s recycle delay, 512×1024 data matrix, 90° shifted sine-bell functions in t_1 and t_2 , zero filled to a final data matrix of 1024×1024 , 3890 Hz sweep width. The respective cross-sections for the individual sugars and the assigned signals are indicated.

ms). This assignment was supported by semiquantitative information the coupling constants obtained from cross-sections of a resolution-enhanced TOCSY spectrum with a digital resolution of 2.7 Hz/point, and 1D-TOCSY spectra which showed large vicinal couplings for all ring protons. The ^1H NMR data are shown in Table 4.

Table 4

¹H NMR data (δ in ppm)^a for the saccharides of J₁

Residue	1	2	3	4	5	6
α -rhap (A)	5.880	5.165	4.915	4.455	4.520	1.760
β -6-deoxy-glc p (B)	4.380	3.990	4.190	5.340	3.670	1.340
β -6-deoxy-glc p (C)	4.815	3.960	4.090	3.690	3.670	1.590
β -glc p (D)	6.030	3.990	4.150	4.170	3.930	4.315
						4.190
β -xyl p (E)	4.980	4.020	4.040	4.120	4.390	
					4.590	
α -ara p (F)	5.140	5.520	4.380	4.385	4.290	
					3.740	
β -glc p (G)	5.310	3.970	4.130	4.055	4.940	4.480
						4.180
β -glc p (H)	4.880	4.010	4.170	4.115	4.060	4.630
						4.220
α -araf (I)	6.250	4.970	4.785	4.725	4.240	
					4.130	

^a Chemical shifts with the highest field signal of C₅D₅N as reference (δ 7.19 ppm).

We now proceed to assign ¹³C resonances of residue G. In principle, ¹³C resonances can be assigned easily by direct correlation in CH-COSY or HMQC spectra using the assigned proton resonances. But in our case, only ¹³C resonances of G₁ (105.73 ppm), G₄ (71.79 ppm), G₅ (78.14 ppm) and G₆ (62.76 ppm) were assigned by direct correlation with the respectively assigned protons. The assignment of G₂ and G₃ was not possible owing to severe signal overlap in both ¹³C and ¹H dimensions. The problem was resolved by use of HMQC-COSY and HMQC-TOCSY spectra which are useful for carbohydrates in which strong coupling and overlapping peaks in ¹H and ¹³C spectra pose difficulties. The ¹³C resonance at 75.40 ppm was assigned to G₂ by correlation with the anomeric proton in the HMQC-COSY spectrum (Table 5). In the HMQC-TOCSY spectrum correlation peaks from the anomeric proton can be traced to all ¹³C resonances of residue G, so the remaining unassigned resonance at 78.39 ppm was assigned to G₃. All correlations from HMQC-COSY and HMQC-TOCSY spectra are summarized in Table 5, and the ¹³C assignments are listed in Table 6.

Residue H.—Like residue G, residue H was also assigned to a β -glucopyranoside. The anomeric proton of residue H shows correlation peaks to all its ring protons in TOCSY spectra (τ_m = 60 and 120 ms). Only the ¹³C assignment of H₁, H₄ and H₆ was established by direct correlation in the CH-COSY spectrum. Difficulty was encountered for residue H due to the signal overlap as noted above for residue G. The ¹³C resonances were assigned with assistance from a HMQC-COSY spectrum which gave the unambiguous assignment of carbon H₂ by correlation with the anomeric proton, and the HMQC-TOCSY spectrum which showed correlations between anomeric proton and ring carbons H₃ and H₅ although they overlapped with other signals (Tables 5 and 6). The β -configuration was identified by ¹J_{CH} (157 Hz) and ³J_{H₁,H₂} (7.9 Hz) data, and confirmed by cross-peaks between the anomeric proton and protons H-H₃ and H-H₅ in the NOESY spectrum.

Table 5

Connectivities observed in HMQC-COSY, HMQC-TOCSY and HMBC spectra for the saccharides of J₁

¹ H signal	Connectivities (¹³ C signals)			
	HMQC-COSY	HMQC-TOCSY	HMBC	
A ₁		A ₂	A ₃ , A ₅ ,	D ₂ ^a
A ₂	A ₃	A ₁ , A ₃		
A ₃		A ₂ , A ₄		
A ₄	A ₃	A ₃ , A ₅	A ₃ , A ₅ , A ₆ ,	I ₁ ^a
A ₅	A ₄	A ₃ , A ₄	A ₄	
B ₁	B ₂	B ₂ , B ₃ , B ₄ , B ₅		X ₆ ^a
B ₂		B ₁ , B ₃ , B ₄ , B ₅	B ₁ , B ₃	
B ₃		B ₁ , B ₂ , B ₄ , B ₅	B ₁	
B ₄		B ₁ , B ₂ , B ₃ , B ₅	B ₃ , B ₅ , B ₆ ,	T ₁ ^a
B ₅		B ₁ , B ₂ , B ₃ , B ₄		
C ₁	C ₂	C ₂ , C ₃ , C ₄ , C ₅	C ₂ , C ₃ ,	T ₆ ^a
C ₂		C ₁ , C ₃ , C ₄ , C ₅	C ₁	
C ₃		C ₁ , C ₂ , C ₄ , C ₅	C ₄	
C ₄	C ₃ , C ₅	C ₁ , C ₂ , C ₃ , C ₅	C ₂ , C ₃ , C ₅	
C ₅	C ₄	C ₁ , C ₂ , C ₃ , C ₄		
D ₁	D ₂	D ₂ , D ₃ , D ₄ , D ₅		28 ^a
D ₂	D ₁	D ₁ , D ₃	D ₄	
D ₃		D ₁ , D ₂ , D ₅		
D ₄	D ₅	D ₁ , D ₃ , D ₅		
D ₅	D ₄	D ₁ , D ₄		
E ₁	E ₂	E ₂ , E ₃ , E ₄ , E ₅		F ₂ ^a
E ₂	E ₁	E ₁ , E ₃ , E ₄	E ₃	
E ₃		E ₁ , E ₂ , E ₄ , E ₅	E ₂ , E ₄	
E ₄		E ₁ , E ₂ , E ₃ , E ₅		
E ₅		E ₁ , E ₃ , E ₄	E ₃ , E ₄	
E ₅	E ₄	E ₁ , E ₃ , E ₄	E ₄	
F ₁	F ₂	F ₂ , F ₃ , F ₄	F ₃ , F ₅ ,	H ₆ ^a
F ₂		F ₁ , F ₃ , F ₅	F ₁ , F ₃ , F ₄ ,	E ₁ ^a
F ₃		F ₁ , F ₂ , F ₄		
F ₄	F ₅	F ₁ , F ₃ , F ₅		
F ₅		F ₄	F ₁ , F ₃ , F ₄	
F ₅		F ₄	F ₁	
G ₁	G ₂	G ₂ , G ₃ , G ₄ , G ₅ , G ₆		A ₃ ^a
G ₂		G ₁	G ₁ , G ₃	
G ₃	G ₄	G ₁	G ₄	
G ₄		G ₁	G ₆	
G ₅	G ₆	G ₁		
H ₁	H ₂	H ₂ , H ₃ , H ₄ , H ₅		3 ^a
H ₂		H ₁ , H ₄	H ₁	
H ₃		H ₁ , H ₄	H ₄	
H ₄		H ₆		
H ₅	H ₄	H ₁ , H ₄ , H ₆		
I ₁	I ₂	I ₂	I ₄ ,	A ₄ ^a
I ₂	I ₃	I ₁ , I ₃ , I ₄ , I ₅	I ₁ , I ₃	
I ₃	I ₄	I ₂ , I ₄ , I ₅	I ₂ , I ₅	
I ₄	I ₃	I ₂ , I ₃ , I ₅		
I ₅		I ₂ , I ₃ , I ₄		
I ₅	I ₄	I ₂		
21				X ₁ ^a

^a Cross-peaks indicate the linkage positions.

Table 6

 ^{13}C NMR data (δ in ppm)^a for the saccharides of J₁

Residue	1	2	3	4	5	6
α -rha <i>p</i> (A)	101.76 (168) ^b	70.53	82.03	78.93	69.15	18.81
β -6-deoxy-glc <i>p</i> (B)	99.29 (160)	75.59	75.59	77.15	70.17	17.09
β -6-deoxy-glc <i>p</i> (C)	99.19 (160)	75.40	78.39	76.82	72.64	18.81
β -glc <i>p</i> (D)	95.67 (161)	76.82	78.04	71.22	79.06	61.95
β -xyl <i>p</i> (E)	106.21 (148)	75.40	77.87	70.83	67.16	
α -ara <i>p</i> (F)	102.22 (166)	80.36	72.53	67.39	64.20	
β -glc <i>p</i> (G)	105.73 (160)	75.40	78.39	71.79	78.14	62.76
β -glc <i>p</i> (H)	106.76 (157)	75.60	78.39	72.22	76.07	69.52
α -ara <i>f</i> (I)	111.02 (173)	84.42	78.39	85.43	62.55	

^a Chemical shifts with the highest field signal of C₅D₅N as reference (δ 123.5 ppm).^b $^1J_{\text{CH}}$ in Hz.

Residue D.—The cross-section of the TOCSY spectrum of residue D through the anomeric proton shows only six signals including H-D₁ and both H-D₆'s, posing a difficulty for the recognition of the residue. It was decided that residue D is a hexopyranose rather than a pentopyranose because the strong coupling of H-D₃ and H-D₄ was identified by iterative comparison of data from CH-COSY, HMQC-COSY and HMQC-TOCSY spectra (Table 5) which show six carbon signals (Table 6). This result is supported by the distorted cross-peak of H-D₁/H-D₃ in the TOCSY spectrum, indicating an overlap of H-D₃ and H-D₄. Residue D was assigned to a β -glucopyranoside on the basis of $^1J_{\text{CH}}$ (160 Hz) and $^3J_{\text{H1,H2}}$ (7.9 Hz), and NOESY cross-peaks of H-D₁/H-D₃ and H-D₁/H-D₅, as discussed above for residue G.

Residue B and C.—Both anomeric protons of residues B and C show correlation peaks with methyl protons H-6 in the TOCSY spectrum (τ_m = 120 ms), indicating a *gluco* configuration. Meanwhile, the $^3J_{\text{H1,H2}}$ values for residues B and C were 7.1 and 7.5 Hz, respectively, excluding the *rhamno* configuration and indicating a β -configuration which was supported by $^1J_{\text{CH}}$ values (160 Hz for both). Therefore, residues B and C were assigned to 6-deoxy- β -glucopyranosides. The procedure used to assign ^{13}C resonances was the same as discussed above (Table 5). The resulting ^1H and ^{13}C data are shown in Tables 4 and 6.

Residue A.—The anomeric proton of residue A shows only correlation with A₂ in the TOCSY spectrum (τ_m = 60 ms), but the correlation can be traced from methyl proton H-A₆ to H-A₂ and to H-A₁ in the same spectrum. Along with $^1J_{\text{CH}}$ (168 Hz) and semiquantitative $^3J_{\text{H1,H2}}$ and $^3J_{\text{H2,H3}}$ (both < 5 Hz) from the cross-sections of the high-res-

olution TOCSY spectrum, residue A was assigned to α -rhamnopyranoside. The ^{13}C assignments of residue A were obtained from the CH-COSY spectrum because of the well-separated ^1H and ^{13}C signals (Tables 4 and 6).

Residue E.—The cross-section of the TOCSY spectrum of residue E through the anomeric proton shows six signals including H-E₁ and both H-E₅'s, and five carbons were observed in the HMQC-TOCSY spectrum (Table 5) in which the anomeric proton shows connectivities with carbons E₂, E₃, E₄ and E₅. The pyranoside pattern was confirmed by three-bond intra-ring correlations between the anomeric carbon E₁ and H-E₅'s in the HMBC spectrum (Fig. 2). The β -configuration was assigned by $^1J_{\text{CH}}$ (148 Hz) and $^3J_{\text{H1,H2}}$ (7.1 Hz), and NOESY cross-peaks of H-E₁/H-E₃ and H-E₁/H-E₅. Thus, residue E was assigned to a β -xylopyranoside. The ^1H and ^{13}C data are listed in Tables 4 and 6.

Residue F.—The anomeric proton of residue F shows correlations only with H-F₂ and H-F₃ (4.380 ppm) in the TOCSY spectrum ($\tau_m = 60$ ms), indicating the existence of a small $^3J_{\text{H3,H4}}$ value. Cross-section through the H-F₅' shows correlation with the same signal at 4.380 ppm which was assigned to H-F₄ and H-F₃. The strong coupling between H-F₃ and H-F₄ was identified by the observation of five carbon signals from the HMQC-TOCSY spectrum (Table 5) in which the anomeric proton shows connectivities with carbons F₂, F₃ and F₄. Two carbons F₃ and F₄ show overlapping cross-peaks with the same proton signal at 4.380 ppm in the CH-COSY spectrum. The pyranoside pattern was identified by the cross-peak between anomeric carbon F₁ and proton H-F₅'s observed in the HMBC spectrum (Fig. 2). The α -anomeric configuration was established by $^1J_{\text{CH}}$ (166 Hz) and $^3J_{\text{H1,H2}}$ (4.2 Hz) values. Residue F was then assigned to an α -arabinopyranoside.

Residue I.—Six proton resonances, including both H-I₅'s, were identified in the TOCSY spectrum (Fig. 1), and their assignments were made following the correlations in the COSY spectrum. The ^{13}C assignments in residue I were obtained from the CH-COSY spectrum, and the result was confirmed by information from the HMQC-TOCSY spectrum (Table 5). The relative downfield carbon resonances I₁ (111.02 ppm), I₂ (84.42 ppm) and I₃ (78.39 ppm) suggest a furanoside structure, which was established by the observed cross-peak between the anomeric proton and carbon I₄ in the HMBC spectrum.

The NMR strategy discussed above is not readily applied to a furanoside. The assignment of residue I to an α -arabinofuranoside was indicated by comparison of the ^{13}C assignments with literature values [8] which shows the similar ^{13}C chemical shifts for an α -L-arabinofuranoside residue at the non-reducing end. The complete assignments of ^1H and ^{13}C spectra are summarized in Tables 4 and 6.

Determination of linkages.—We now proceed to identify linkage positions from HMBC cross-peaks. The data in Fig. 2 and Table 5 illustrate that all the anomeric protons show correlations with carbon atoms across the glycosidic linkage through three-bond coupling. The large number of intra-ring cross-peaks observed in the HMBC spectrum support the ^1H and ^{13}C assignments discussed above. The anomeric proton of residue E was correlated with carbon F₂, and the H-F₁ was correlated with carbon H₆, giving a trisaccharide segment. The ^{13}C assignment of the trisaccharide (Table 6) was in good agreement with data on the same segment found in the literature [3], supporting the

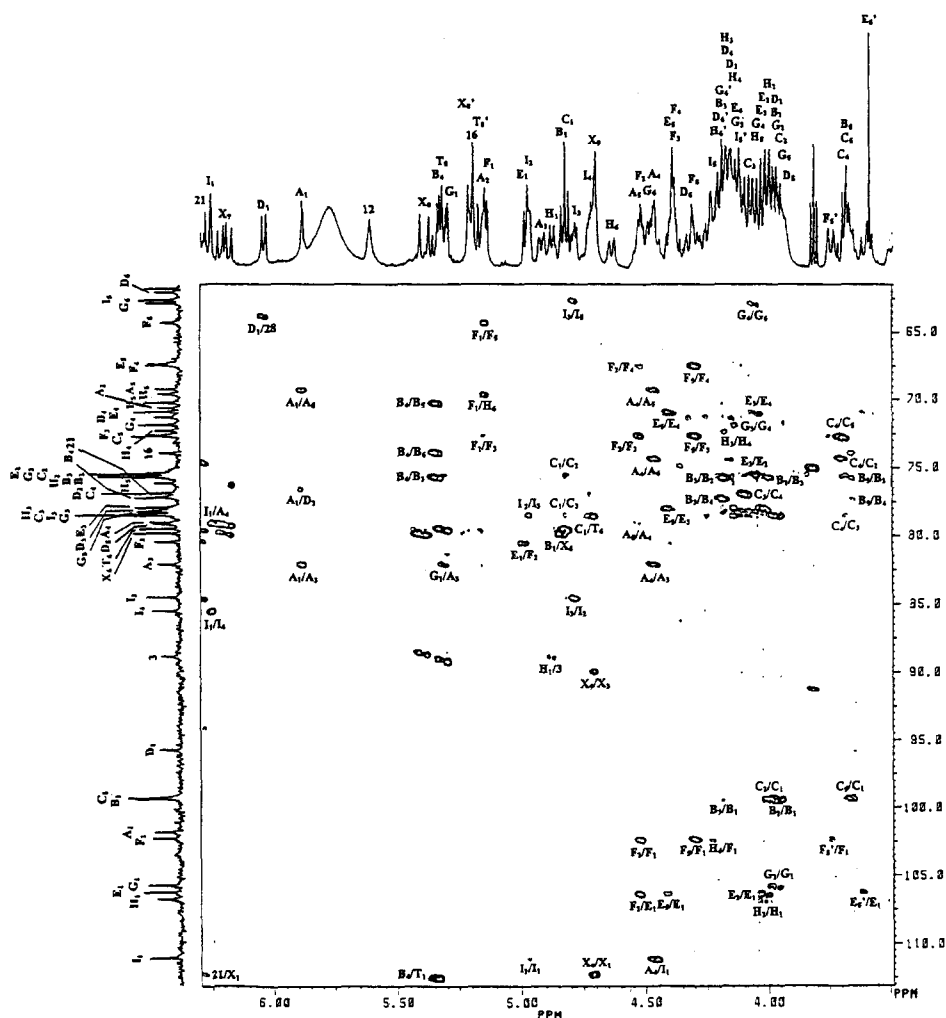


Fig. 2. HMBC spectrum of regions F_1 , 113–62 ppm (^{13}C), and F_2 , 6.3–3.4 ppm (^1H), 1 s recycle delay, 256×1024 data matrix, Δ_1 and Δ_2 durations of 3.4 and 70 ms, respectively, zero filled to a final data matrix of 512×1024 , sweep widths of 2777 Hz in t_2 and 6962 Hz in t_1 . The respective connectivities are indicated with proton resonances (first) and carbon resonances (second). Cross-peaks of $D_1/28$, $21/X_1$ and B_4/T_1 are folded in the F_1 dimension (^{13}C). Cross-peak of A_1/D_2 is observed at lower level.

result we obtained. The trisaccharide EFH is linked to carbon 3 of the triterpene as indicated by a cross-peak between the anomeric proton of residue H and carbon 3 of the triterpene. The triterpene proton 21 shows long-range correlation with the monoterpene carbonyl X_1 , indicating the linkage position. Another monoterpene is linked to B_4 as indicated by a long-range cross-peak $H-B_4/T_1$ in the HMBC spectrum. The monoterpene X is linked to residue B as indicated by a cross-peak $H-B_1/X_6$, and monoterpene

T to residue C from cross-peak H-C₁/T₆. The tetrasaccharide segment was established via a long-range cross-peak across the glycosidic linkage, and the anomeric proton of residue D is linked to carbon 28 of triterpene via cross-peak H-D₁/28. The anomeric proton resonance of residue A shows a correlation peak in HMBC with overlapping carbon signals of C₄ and D₂ at 76.82 ppm, precluding an accurate assignment of the linkage. The glycosidic linkage between residue A and D was identified by cross-peak H-A₁/H-D₂ observed in the NOESY spectrum. These glycosidic linkage results also explain the glycosylation downfield effect on ¹³C chemical shifts at the linkage positions.

The discussion above demonstrates complete assignment of the ¹H and ¹³C spectra of a complex carbohydrate and that HMBC data are sufficient to identify glycosidic linkages. Although the nine sugars in the triterpenoid pose a relatively difficult problem in structure determination, having different anomeric configurations and ring forms, the complete resonance assignments and structure determination were obtained because of the redundancy provided by the different NMR methods used. If difficulties in the interpretation are encountered due to small *J* values or to unfortunate overlap of resonances, there are often alternative methods to be chosen.

3. Experimental

Isolation of Julibroside J₁.—The crude saponin fraction was chromatographed on silica gel columns and gradient eluted with CHCl₃ and CHCl₃:MeOH = 100:1 to 1:1. The major fraction was subjected to gel filtration on Sephadex LH-20 with MeOH, and purified by preparative HPLC with MeOH:H₂O = 62:38 to afford J₁ as a white powder, mp 170–172 °C. Anal. Calcd for C₁₀₁H₁₆₀O₄₉ · 3½H₂O: C, 54.61; H, 7.52. Found: C, 54.18; H, 6.93.

General methods.—HPLC was carried out on a μ-Bondapak C-18 (Waters 510) column.

NMR spectroscopy.—The sample (15 mg) was dissolved in 99.5% C₅D₅N (0.5 mL). Spectra were recorded on a Bruker AM-500 spectrometer equipped with an Aspect 3000 computer. The upfield peak of the three solvent signals in ¹H and ¹³C spectra were taken as the internal standard (δ 7.19 ppm for ¹H and δ 123.5 ppm for ¹³C). All experiments were carried out at 30 °C. The following techniques and parameters were used. COSY [14], DQF-COSY [15] and PH-NOESY [16] were performed with 1, 2 and 3 s recycle delays, respectively; TOCSY (1) [17]: 2 s recycle delay, 43 μs 90° pulse, 120 ms spin-lock mixing time using the low power transmitter (TLO mode); TOCSY (2): 1.5 s recycle delay, 32 μs 90° pulse in reverse mode, 60 ms spin-lock mixing time. CH-COSY [18], HMQC-COSY [19] and HMQC-TOCSY [19] were performed with a 1 s recycle delay and a fixed delay of 3.4 ms (1/2 ¹J_{CH}). A spin-lock of 80 ms was used for the HMQC-TOCSY; HMBC (1) [20]: 1 s recycle delay, Δ₁ and Δ₂ duration of 3.4 and 70 ms, respectively; HMBC (2): the chemical shifts were folded in two dimensions.

References

- [1] T. Konoshima, M. Kozuka, T. Sawada, and T. Kimura, *Chem. Pharm. Bull.*, 35 (1987) 46–52.
- [2] T. Konoshima, T. Sawada, and T. Kimura, *Chem. Pharm. Bull.*, 33 (1985) 4732–4739.
- [3] T. Konoshima, and T. Sawada, *Chem. Pharm. Bull.*, 30 (1982) 4082–4087.
- [4] T. Konoshima, M. Kozuka, and T. Kimura, *Chem. Pharm. Bull.*, 35 (1987) 1982–1990.
- [5] T. Konoshima, T. Sawada, and T. Kimura, *Chem. Pharm. Bull.*, 32 (1984) 4833–4841.
- [6] T. Konoshima and T. Sawada, *Chem. Pharm. Bull.*, 32 (1984) 2617–2621.
- [7] J. Kinjo, K. Araki, K. Fukui, H. Higuchi, T. Ikeda, T. Nohara, Y. Ida, N. Takemoto, M. Miyakoshi, and J. Shoji, *Chem. Pharm. Bull.*, 40 (1992) 3269–3273.
- [8] R. Encarnacion, L. Kenne, G. Samuelsson, and F. Sandberg, *Phytochemistry*, 20 (1981) 1939–1942.
- [9] C. Abeygunawardana, C.A. Bush, and J.O. Cisar, *Biochemistry*, 29 (1991) 234–248.
- [10] C. Abeygunawardana and C.A. Bush, *Adv. Biophys. Chem.*, 3 (1993) 199–249.
- [11] S.W. Homans, *Progr. Nucl. Magn. Reson. Spectrosc.*, 22 (1990) 55–81.
- [12] R. Kasai, M. Okihara, J. Asakawa, K. Mizutani, and O. Tanaka, *Tetrahedron*, 35 (1979) 1427–1432.
- [13] C. Altona and C.A.G. Haasnoot, *Org. Magn. Reson.*, 13 (1980) 417–429.
- [14] K. Nagayama, A. Kumar, K. Wüthrich, and R.R. Ernst, *J. Magn. Reson.*, 40 (1980) 321–334.
- [15] A.J. Shika and R. Freeman, *J. Magn. Reson.*, 51 (1983) 169–173.
- [16] G. Bodenhausen, H. Kogler, and R.R. Ernst, *J. Magn. Reson.*, 58 (1984) 370.
- [17] A. Bax and D.G. Davis, *J. Magn. Reson.*, 65 (1985) 355.
- [18] A. Bax and G. Morris, *J. Magn. Reson.*, 42 (1981) 501.
- [19] L. Lerner and A. Bax, *J. Magn. Reson.*, 69 (1986) 375–380.
- [20] A. Bax and M.F. Summers, *J. Am. Chem. Soc.*, 108 (1986) 2093–2094.